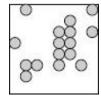
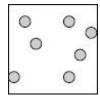
Science 5 - Sawyer Matter [Exam ID:6291]

1 Which of the diagrams best shows the arrangement of molecules in a solid?

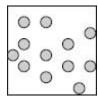
A



B



 \mathbf{C}

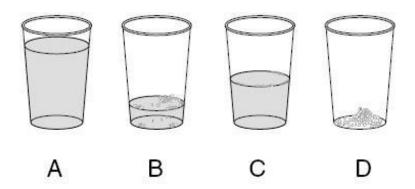


 \mathbf{D}



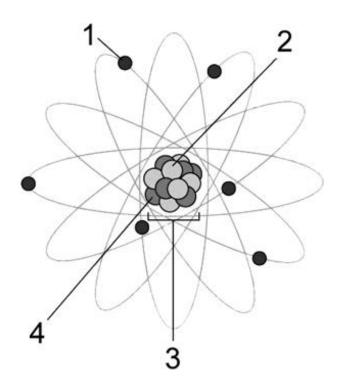
2 Water, ice, and steam are alike because they —

- ${\bf F}$ are the same compound
- G have the same shape
- H look the same
- J feel the same



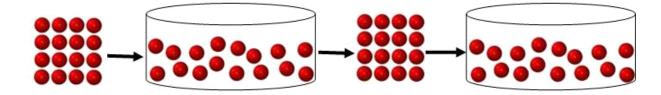
Which set of pictures shows what happens to a glass of salt water when it is left out on a counter for several weeks?

- $A \qquad \mathsf{D} \,\to\, \mathsf{B} \,\to\, \mathsf{A} \,\to\, \mathsf{C}$
- $B \qquad \mathsf{A} \,\to\, \mathsf{C} \,\to\, \mathsf{B} \,\to\, \mathsf{D}$
- $C \quad \mathsf{C} \to \mathsf{D} \to \mathsf{A} \to \mathsf{B}$
- \mathbf{D} $\mathsf{B} \to \mathsf{A} \to \mathsf{C} \to \mathsf{D}$



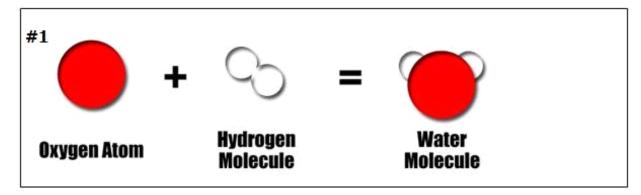
Look at the atom. Section 2 shows which part of the atom?

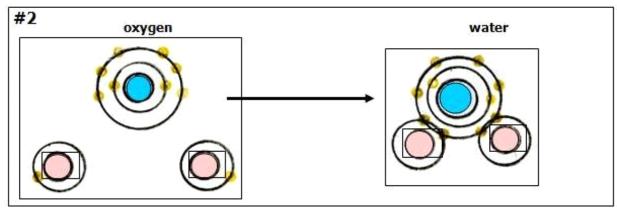
- F Nucleon
- **G** Proton
- H Electron
- J Tritron



This diagram shows two states of matter changing from one to another. Which states of matter and their changes are taking place?

- A Gas to a liquid, liquid to a gas, gas to a liquid
- B Solid to a liquid, liquid to a solid, solid to a liquid
- C Solid to a gas, gas to a solid, solid to a gas
- **D** Liquid to a gas, gas to a liquid, liquid to a gas

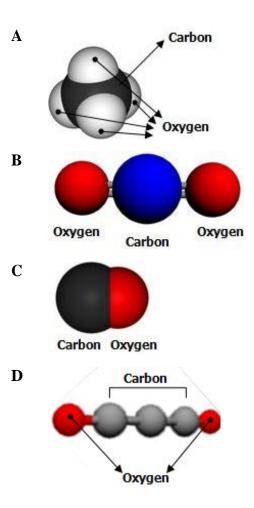




These two pictures show how hydrogen and oxygen combine to form water. Based on these pictures, which of these is a good prediction for the future?

- F Helium and oxygen combine to make water.
- G Water will always look like this
- H Combining hydrogen and water will make oxygen.
- J Water will always look like this .

Carbon dioxide (CO₂) is a compound. Which of these is an accurate model of this compound?



8 Which of these is a way that elements and compounds are related?

- F Elements can form only one kind of compound.
- G Elements always combine in the same way to form compounds.
- H Elements combine in many different ways to form different compounds.
- J Compounds combine to form elements.

9 Which of these is NOT a compound?

- A NaCI
- B Au
- C CO
- **D** H₂O

10 Manuel observed steam rising from the pot of boiling water. The molecules in the steam are —

- F packed tightly and barely moving
- G moving very slowly as they rise
- H spread apart and moving very fast
- J packed tightly and moving a little

11 What determines whether matter is a solid, liquid, or gas?

- A Number of elements
- **B** Number of electrons
- C Speed of atoms or molecules
- **D** Number of neutrons

12 What orbits the nucleus of an atom? F Nucleus \mathbf{G} Electrons Η **Protons** J Neutrons 13 Atoms are the smallest part of a(n) — A electron В proton \mathbf{C} element D neutron

protons and neutrons

neutrons and electrons

protons and electrons

mixtures and compounds

What is the smallest part of matter?

The nucleus in the center of the atom is made up of —

14

15

F

G

H

J

A

В

 \mathbf{C}

D

Voltage

An atom

A circuit

A switch

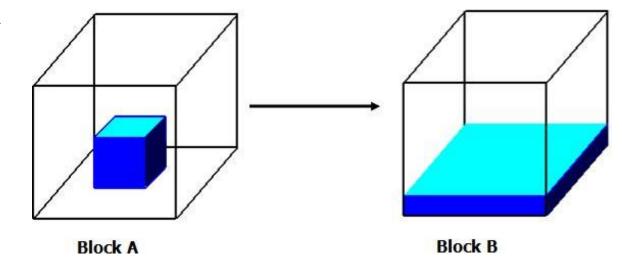
| | fruit salad is an example of a — | |
|----|----------------------------------|---|
| | F | proton |
| | G | compound |
| | H | mixture |
| | J | solution |
| | | |
| 17 | | gar is poured into a glass of water and dissolves. This is an example a(n)— |
| | A | ratio |
| | В | compound |
| | C | element |
| | D | solution |
| | | |
| 18 | Wł | nich is a solution? |
| | F | Mixed fruit salad |
| | G | Lettuce and cucumber salad |
| | H | Peanuts and fruit |
| | J | Chocolate milk |
| | | |

Marcella served a fruit salad made with bananas, grapes, and apples. The

16

19 How are mixtures and solutions the alike?

- A They are both solids.
- B They dissolve in water.
- C They are both easy to separate.
- **D** They have two or more ingredients.
- Joe combined sand and iron filings in a cup. He separated them using a magnet. Before he separated them, they were an example of -
 - F a mixture
 - G a solution
 - H a neutron
 - J a molecule



Each of these cubes represent a state of matter. What would need to happen to make the change from Block A to Block B?

- A Temperature would need to increase.
- B Temperature would need to stay the same.
- C Temperature is not a factor.
- **D** Temperature would need to decrease.

Which of these is a mixture?

- F Lemonade
- **G** Sugar
- H Water
- J Salt

| 23 | A student makes a fruit drink by stirring a powdered mix into cold water. |
|----|---|
| | Why is the fruit drink a solution? |

- A The powder dissolves in the water.
- B The water is the proper temperature.
- C The student stirs the water.
- **D** The water changes color.

Juan made a cake using eggs, flour, milk, and sugar. A baked cake is an best example of a —

- F dry mixture
- G liquid phase
- H solution
- J electron

25 All of the following are considered liquids, but...

- A Kool-aid
- **B** Coca-Cola
- C water
- D peanut butter

Which of these will change solid iron to a liquid?

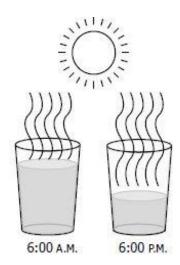
- F Increasing its temperature
- G Adding water to the iron
- **H** Crushing the solid iron
- J Raising the air pressure

What causes water molecules to change from a liquid into a gas?

- A Drinking the water
- **B** Heating the water
- C Pouring the water
- **D** Cooling the water

Which of the following will *not* happen when heat is applied?

- F Water freezes and becomes solid ice.
- G Liquid water boils and becomes steam or gas.
- H Liquid water evaporates and becomes a gas.
- J Solid ice melts and becomes a liquid.



The process shown would be classified as —

- A precipitation
- B transpiration
- C condensation
- **D** evaporation



What is happening to the water molecules in this bottle of water?

- **F** They are condensing.
- G They are freezing.
- **H** They are precipitating.
- **J** They are evaporating.

Which of these shows how frozen water changes as the temperature of the air increases?

- A Solid \rightarrow liquid \rightarrow gas
- $B \qquad \text{Gas} \, \rightarrow \, \text{liquid} \, \rightarrow \, \text{solid}$
- C Liquid \rightarrow gas \rightarrow solid
- \mathbf{D} Solid \rightarrow gas \rightarrow liquid

32 When ice cream is left out of a freezer, the ice cream changes from a -

- F gas to a liquid
- G solid to a gas
- H liquid to a gas
- ${f J}$ solid to a liquid

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